Mindluster Platform

About the course Chemistry

What are the 5 types of chemistry? Traditionally, chemistry has been broken into five main subdisciplines: Organic, Analytical, Physical, Inorganic and Biochemistry. What is chemistry in simple words? The definition of chemistry is a branch of science that deals with the form and properties of matter and substances or the interaction between individuals. An example of chemistry is the study of protons and neutrons. An example of chemistry is the feeling of affection and attraction between a couple. noun. How do you define chemistry? Chemistry, the science that deals with the properties, composition, and structure of substances (defined as elements and compounds), the transformations they undergo, and the energy that is released or absorbed during these processes. What are the 3 laws of chemistry? Laws of Chemistry The law of the conservation of mass. The law of constant proportions. The law of multiple proportions. The law of reciprocal proportions. Which branch of chemistry is best? Organic chemistry offers the maximum job opportunities. Though the industries of all these branches are different. Like if you have done specialization in physical chemistry than you should go for detergents, polymer based industries. Why is chemistry so important? Chemistry is important because everything you do is chemistry! Even your body is made of chemicals. Chemical reactions occur when you breathe, eat, or just sit there reading. All matter is made of chemicals, so the importance of chemistry is that it's the study of everything What is chemistry and its uses? Chemistry is a fundamental and enabling science that investigates moleculesâ€”the building blocks of all matterâ€”and how they interact to affect the composition, structure and properties of substances. What are the characteristics of chemistry? Chemistry is the scientific discipline involved with elements and compounds composed of atoms, molecules and ions: their composition, structure, properties, behavior and the changes they undergo during a reaction with other substances.

Science Category's Courses

Course Lesson(47)

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Lesson 2 : **The Nucleus Crash Course Chemistry 1**

Lesson 3 : **Unit Conversion Significant Figures Crash Course Chemistry 2**

Lesson 4 : **The Creation of Chemistry The Fundamental Laws Crash Course Chemistry 3**

Lesson 5 : **The Periodic Table Crash Course Chemistry 4**

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Lesson 7 : **Stoichiometry Chemistry for Massive Creatures Crash Course Chemistry 6**

Lesson 8 : **Water Solutions for Dirty Laundry Crash Course Chemistry 7**

Lesson 9 : **Acid Base Reactions in Solution Crash Course Chemistry 8**

Lesson 10 : **Precipitation Reactions Crash Course Chemistry 9**

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Lesson 34 : **Doing Solids Crash Course Chemistry 33**

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Lesson 47 : **The Global Carbon Cycle Crash Course Chemistry 46**

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